# Magnesium Metal Ribbon

MSDS # 426.10

Scholar

HMIS (0 to 4)

Reactivity

Section 1:

# **Product and Company Identification**

# Magnesium Metal Ribbon

Synonyms/General Names: N/A Product Use: For educational use only

Manufacturer: Columbus Chemical Industries, Inc., Columbus, WI 53925.

24 Hour Emergency Information Telephone Numbers CHEMTREC (USA): 800-424-9300

CANUTEC (Canada): 613-424-6666 ScholAR Chemistry; 5100 W. Henrietta Rd, Rochester, NY 14586; (866) 260-0501; www.Scholarchemistry.com

Section 2:

## Hazards Identification

Silver metal ribbon, no odor

WARNING! Flammable solid, dangerous when wet.

Flammable solid, keep away from all ignition sources. Contact with water produces flammable

Target organs: Skin, eyes and respiratory system.

This material is considered hazardous by the OSHA Hazard Communication Standard (29 CFR 1910.1200).

#### Section 3:

# Composition / Information on Ingredients

Magnesium (7439-95-4), >99%

# Section 4:

#### **First Aid Measures**

Always seek professional medical attention after first aid measures are provided.

Eyes: Skin:

Immediately flush eyes with excess water for 15 minutes, lifting lower and upper eyelids occasionally. Immediately flush skin with excess water for 15 minutes while removing contaminated clothing.

Ingestion:

Call Poison Control immediately. Rinse mouth with cold water. Give victim 1-2 tbsp of activated charcoal mixed

with 8 oz water.

Inhalation:

Remove to fresh air. If not breathing, give artificial respiration.

### Section 5:

## **Fire Fighting Measures**

Flammable solid. When heated to decomposition, emits acrid fumes

Protective equipment and precautions for firefighters: Do Not Use carbon dioxide, foam, water or halogenated extinguishing agents. Use class D extinguisher or smother with dry sand, dry clay, dry ground limestone or dry graphite. Firefighters should wear full fire fighting turn-out gear and respiratory protection (SCBA). Material is not sensitive to mechanical impact or static discharge.



### Section 6:

### Accidental Release Measures

Use personal protection recommended in Section 8. Isolate the hazard area and deny entry to unnecessary and unprotected personnel. Remove all ignition sources and ventilate area. Sweep up spill and place material in a dry container for disposal. See Section 13 for disposal information.

### Section 7:

### Handling and Storage

Red

Handling: Use with adequate ventilation and do not breathe dust or vapor. Avoid contact with skin, eyes, or clothing. Wash hands thoroughly after handling.

Storage: Store in Flammable Area [Red Storage] with other flammable materials and away from any strong oxidizers. Store in a dedicated flammables cabinet. Store in a cool, dry, well-ventilated, locked store room away from incompatible materials.

#### Section 8:

### **Exposure Controls / Personal Protection**

Use ventilation to keep airborne concentrations below exposure limits. Have approved eyewash facility, safety shower, and fire extinguishers readily available. Wear chemical splash goggles and chemical resistant clothing such as gloves and aprons. Wash hands thoroughly after handling material and before eating or drinking. Use NIOSH-approved respirator with a dust cartridge. Exposure guidelines: Magnesium: OSHA PEL: N/A and ACGIH TLV: N/A, STEL: N/A.

Section 9: Physical and Chemical Properties			
Molecular formula Molecular weight Specific Gravity Vapor Density (air=1) Melting Point Boiling Point/Range Vapor Pressure (20°C) Flash Point: Autoignition Temp.:	Mg. 24.31. 1.74 g/mL @ 20°C. N/A. 651°C. 1107°C. N/A. N/A. 473°C (883°F).	Appearance Odor Odor Threshold Solubility Evaporation rate Partition Coefficient pH UEL LEL	Silver metal ribbon.  No odor.  N/A.  Acids.  N/A. (Butyl acetate = 1).  N/A. (log Pow).  N/A.  N/A.  N/A.  N/A.  N/A.

### Section 10:

# Stability and Reactivity

Avoid heat and ignition sources

Stability: Stable under normal conditions of use.

Incompatibility: Water, acids, chlorine, iodine, bromine and oxidizing agents.

Shelf life: Indefinite if stored properly.

### Section 11:

# **Toxicology Information**

Acute Symptoms/Signs of exposure: Eyes: Stinging pain, burns, watering of eyes, inflammation of eyelids and conjunctivitis. Avoid looking at burning magnesium. Skin: Irritation, redness, burns. Powdered metal ignites readily on skin causing burns. Ingestion: Nausea, vomiting and headache. Inhalation: Rapid irregular breathing, headache, burns to mucous membranes. Inhalation of dust or fumes causes metal fume fever.

Chronic Effects: Repeated/prolonged skin contact may cause dryness or rashes.

Sensitization: none expected

Magnesium: LD50 [oral, rat]; Not Available; LC50 [rat]; Not Available; LD50 Dermal [rabbit]; Not Available Material has not been found to be a carcinogen nor produce genetic, reproductive, or developmental effects.

### Section 12:

## **Ecological Information**

Ecotoxicity (aquatic and terrestrial):

Ecological impact has not been determined

## Section 13:

# **Disposal Considerations**

Check with all applicable local, regional, and national laws and regulations. Local regulations may be more stringent than regional or national regulations. Use a licensed chemical waste disposal firm for proper disposal.

## Section 14:

## Transport Information

**DOT Shipping Name: DOT Hazard Class:** 

Magnesium.

Canada TDG: **Hazard Class:**  Magnesium. 4.1, pg III.

Identification Number:

4.1, pg III. UN1869.

**UN Number:** 

UN1869.

#### Section 15:

## Regulatory Information

EINECS: Listed (231-104-6).

WHMIS Canada: B4, B6: Flammable solid, Reactive flammable material.

TSCA: All components are listed or are exempt.

California Proposition 65: Not listed.

The product has been classified in accordance with the hazard criteria of the Controlled Products Regulations and the MSDS contains all the information required by the Controlled Products Regulations.

### Section 16:

### Other Information

## Current Issue Date: December 22, 2011

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